

# ADVANCED MATERIALS

**Supporting Information**

for

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## Supplementary Section

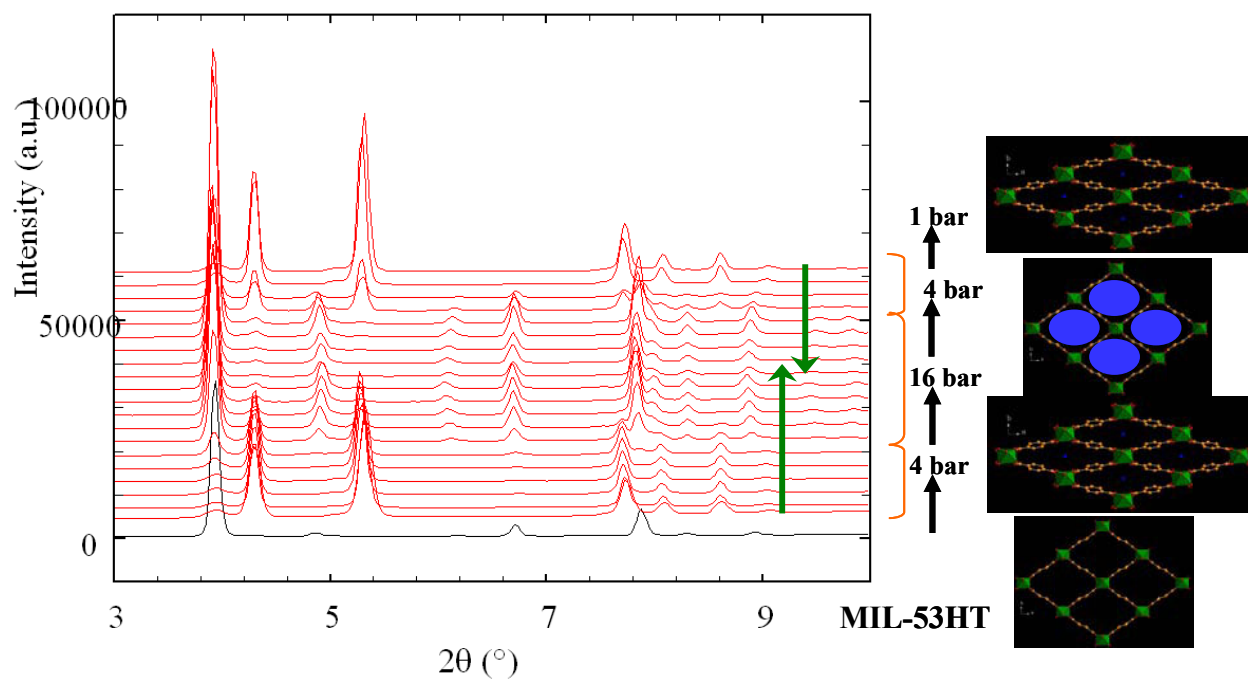
Table S1 : Crystal data for MIL-53LP.

Formula	$[Cr^{III}(OH)(OOC-C_6H_4-COO)].(CO_2)_x (x \sim 0.54)$
Chemical formula	$Cr_4O_{24.32}C_{34.16}H_{20}$
Molar weight (g.mol <sup>-1</sup> )	1020.3
Calculated density (g.cm <sup>-3</sup> )	1.57
Crystal system	Monoclinic
Space group	$C 2/c$ (n°15)
$a$ (Å)	19.713(1)
$b$ (Å)	8.310(1)
$c$ (Å)	6.806(1)
$\beta$ (°)	105.85(1)
$V$ (Å <sup>3</sup> )	1072.48(2)
$Z$	1
Figures of merit	$M_{19}/F_{19}=11/40$
Radiation $\lambda$ (Å)	0.71118
Temperature (K)	195
$2\theta$ range (°)	2-26.7
N. reflections	157
N. independent atoms	10
N. intensity parameters	26
N. profile parameters	9
N. soft distance constraints	24
$R_p$	4.5
$R_F$	12.1
Isotropic thermal factor	1.2(1)
Profile function	Pearson VII
Background	Experimental (36 points)
N. of asymmetry parameters	2

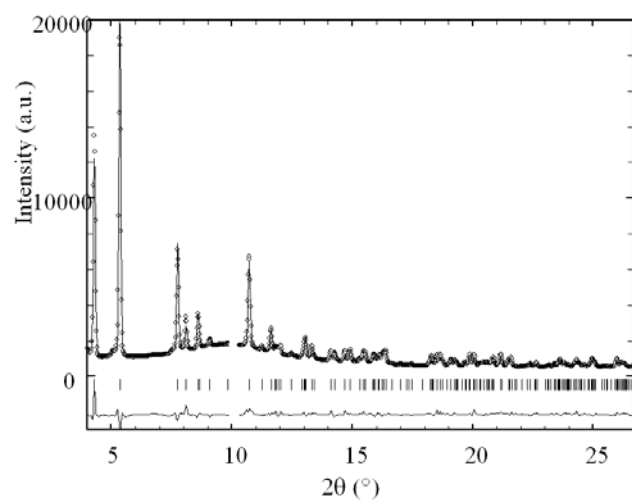
Table S2 : atomic positions of MIL-53LP at 195 K under 1 bar of CO<sub>2</sub>:

(S.G.  $C 1 2/c 1$  (no. 15),  $a = 19.716(1)$  Å,  $b = 8.310(1)$  Å,  $c = 6.805(1)$  Å,  $\beta = 105.85(1)$  °,  $V=1072.6(2)$  Å<sup>3</sup>)

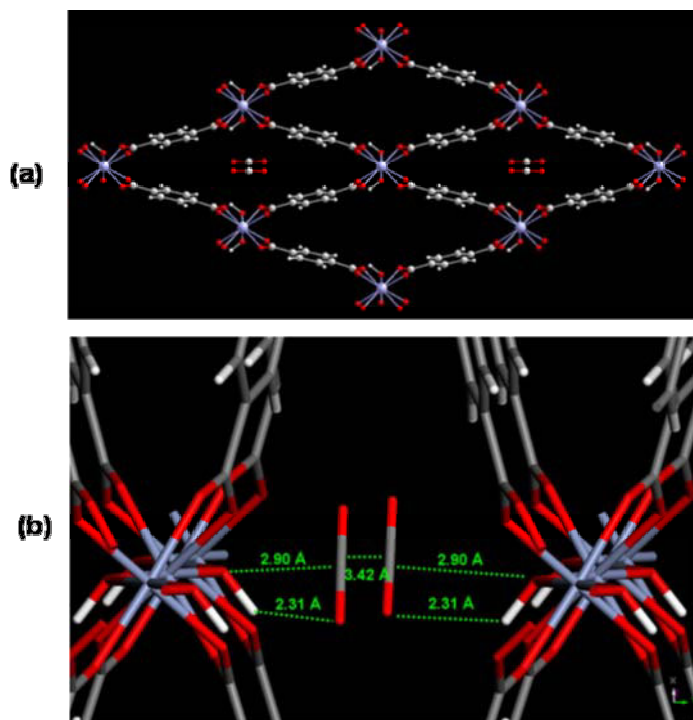
Atom	Wyckoff	Site Occupation	x	y	z
Cr	4a		0	0	0
O(1)	4e		0	-0.106(2)	3/4
O(2)	8f		0.0645(6)	0.165(1)	0.944(2)
O(3)	8f		0.0857(9)	0.120(1)	0.652(2)
C(1)	8f		0.1055(5)	0.169(3)	0.838(2)
C(2)	8f		0.2016(6)	-0.259(3)	0.621(2)
C(3)	8f		0.2756(6)	-0.295(2)	0.707(2)
C(4)	8f		0.179(1)	-0.219(2)	0.419(2)
C	4e	0.54(1)	0	0.557(6)	3/4
O	8f	0.54(1)	0.0469(7)	-0.442(3)	0.920(2)



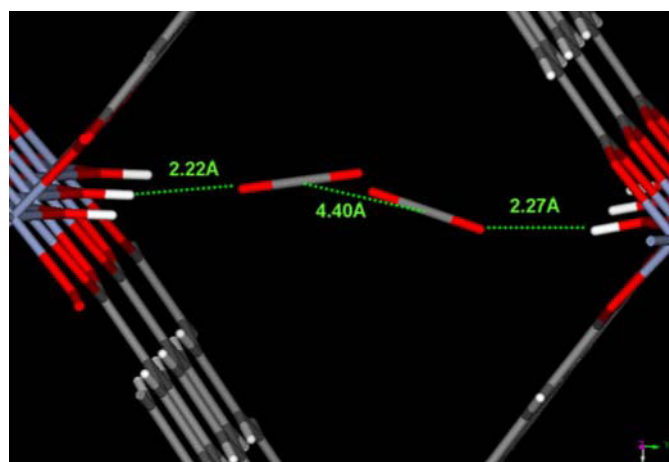
**Figure S1:** left : XRPD patterns of MIL-53(Cr) under CO<sub>2</sub> pressure at 298 K; ( $\lambda=0.71118 \text{ \AA}$ ). A pattern of the outgassed sample, prior to the introduction of CO<sub>2</sub>, is shown in black. Right: schematic view of the various structures of MIL-53 as a function of the pressure.



**Figure S2:** Rietveld fit for MIL-53LP measured at 195 K under 1 bar of CO<sub>2</sub>,  $\lambda=0.71118 \text{ \AA}$ .



**Figure S3:** Top: view of the DFT optimised MIL-53LP structure along the c axis; bottom : view of the •straight-on and direct interactions between CO<sub>2</sub> and the hydroxyl group in the DFT optimised MIL-53LP solid (red circles: oxygen, gray circles: carbon, white circle : hydrogen, blue circle : chromium).



**Figure S4:** view of the •straight-on and direct interactions between CO<sub>2</sub> and the hydroxyl group in the DFT optimised MIL-53HP solid (red circles: oxygen, gray circles: carbon, white circle : hydrogen, blue circle : chromium)

[figS5.pps](#)

**Figure S5:** “movie” of the variation of the MIL53 structure (bands at 1017 and 1022  $\text{cm}^{-1}$ ) and of the  $\nu_2$   $\text{CO}_2$  absorption band during a cycle of  $\text{CO}_2$  adsorption-desorption. 45 spectra recorded in the 0-10 bars pressure range are presented.